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Lewis Acid Catalyzed Intramolecular [3 + 2] Cross Cycloadditions of Cobalt-Alkynylcyclopropane 1,1-Diesters with Carbonyls for Construction of Medium-Sized and Polycyclic Skeletons

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S Supporting Information

[AB](#page-2-0)STRACT: [A Lewis acid](#page-2-0) catalyzed intramolecular $\begin{bmatrix} 3 + 2 \end{bmatrix}$ cross cycloaddition of cobalt-alkynylcyclopropane 1,1-diesters with carbonyls has been successfully developed. Together with simple and efficient postcycloadditions of the cobalt-alkyne moiety, a general and efficient strategy for construction of structurally complex and diverse mediumsized skeletons and related polycycles was supplied successfully.

Developing general and efficient methods for the construction of structurally diverse and complex polycyclic skeletons is important for natural products synthesis and lead discovery. Medium-sized skeletons constitute the basic skeletons of many important biologically active natural products, often embedded within polycycles (Figure 1).¹ These medium-sized skeletons are also useful building blocks in organic synthesis. However, because of entropic and enthal[pi](#page-2-0)c factors, the mediumsized skeletons, and in particular the carbocyclic ones, are much more difficult to be assembled in high efficiency with conventional methods.² Developing efficient and general

Figure 1. Representative natural products with medium-sized ringderived polycyclic carbocycles.

strategies for the construction of medium-sized and related polycyclic skeletons, both in terms of the efficiency for complexity and the generality for diversity, remains a challenging and attractive theme in organic synthesis. $3-6$

We have recently developed an intramolecular cross-cycloaddition (IMCC)^{7d,8} of donor–acceptor [c](#page-2-0)y[cl](#page-3-0)opropanes,⁷ which provided a general and efficient strategy for the construction of medium-sized sk[elet](#page-3-0)ons. To develop a general and [e](#page-3-0)fficient strategy for construction of polycycles deriving from the medium-sized skeletons, one feasible method is to introduce a suitable functional group into the linker of the IMCC. Such a functional group should meet the following criteria: it should configurationally and conformationally favor the cycloaddition, it should activate the cyclopropane as a donor, and it should be applied for further elaboration into structurally diverse skeletons, especially cyclic ones via easy postmodifications.

Alkynyl groups have rich chemistries and can be used as donors (as a propargyl cation). Furthermore, intermolecular [3 + 2] cycloadditions of alkynylcyclopropane 1,1-diester were reported recently by Niggemann, Sakata, and Nishibayashi.^{9d,e} However, this cannot be directly applied in an intramolecular version in which the two reactive sites are difficult to connect. [By](#page-3-0) creatively invoking the Nicholas carbocation intermediate,¹⁰ Christie and Jones et al. described an intermolecular $[3 + 2]$ cycloaddition of cobalt-alkynylcyclopropane (CoACP) 1,[1](#page-3-0) diester with aldehydes or imines under promotion of BF₃· $\mathrm{OEt_2}^{9a,b}$ Kerr et al. reported a Sc $\mathrm{(OTf)}_3$ -catalyzed intermolecular $\left[3+3\right]$ cycloaddition of CoACP 1,1-diester with nitrones.^{9c} We r[easo](#page-3-0)n that the cobalt-alkyne moiety can be thought as a "bent alkyne" to provide an opportunity for the two reactive sit[es](#page-3-0) to meet each other in the intramolecular cycloadditions, which can lead to postfunctionalizations (Scheme 1).

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Scheme 1. $[3 + 2]$ IMCC of CoACP 1,1-Diester for Construction of Structurally Diverse Polycyclic Skeletons

Our initial investigation started from the reaction of substrate 1a for the concept validation. Compound 1a was easily obtained by the methods reported in the literature.^{9a−c} We were delighted to find that most of the Lewis acids could successfully promote the reaction (Table 1) to give the desired [cycl](#page-3-0)oadduct $2a$. The [3]

Table 1. Optimization of the $[3 + 2]$ IMCC Condition ^a				
(OC) ₆ Co ₂	CO ₂ Me CO ₂ Me СНО 1a	conditions	2a	$Co_2(CO)_{6}$ CO ₂ Me CO ₂ Me
entry	catalyst	solvent	time (h)	yield $(\%)^b$
1	SnCl ₄	DCE	$\overline{2}$	86
$\overline{2}$	$BF_3.Et_2O$	DCE	8	27
3	Sn(OTf),	DCE	17	59
$\overline{4}$	Sc(OTf)	DCE	$\overline{2}$	93
5	$Yb(OTf)$ ₃	DCE	17	40
6	Zn(OTf),	DCE	17	12
7	Sc(OTf)	CH ₂ Cl ₂	17	76
8	Sc(OTf)	toluene	17	17
9	Sc(OTf) ₃	DCE	17	53

a
Reaction conditions: 1a (0.1 mmol), catalyst (0.2 equiv), solvent (2.5 mL), rt. b Isolated yields. CUsed 0.1 equiv of Sc(OTf)₃. DCE = 1,2dichloroethane.

 $+ 2$] IMCC proceeded smoothly under catalysis of Sc(OTf)₃ in CH_2ClCH_2Cl (DCE) at room temperature with an excellent yield (93%), which was used as the optimal condition (Table 1, entry 4) for most of the other examples. Structure of 2a was determined by NMR and X-ray crystal analysis.^{11,12}

Under the optimal condition, the scope of the substrates was then examined (Figure 2). $[3+2]$ IMCC of the [tetra](#page-3-0)-substituted cyclopropane 1b also gave the desired cycloadduct 2b successfully. Substrates with either electron-donating (1d and 1g) or electron-withdrawing (1e) substituent on the benzene ring also gave the $[3+2]$ IMCC cycloadducts 2d, 2e, and 2g. The $[3 + 2]$ IMCC of ketone 1f was also successful. When the benzene rings were replaced by cycloalkenes, the $[3 + 2]$ IMCC were successful and gave the corresponding cycloadducts in moderate to good yields (2h−k). These examples are very worthy of attention because the easy introduction 13 of various cyclic rings, and the potential postmodifications of the $C=C$ bridges may provide an efficient and general [st](#page-3-0)rategy for construction of structurally diverse polycyclic skeletons in natural products (Figure 1). The successful $\begin{bmatrix} 3 + 2 \end{bmatrix}$ IMCC of 1l promises to be a potential method for synthesis of the cyclooctane-fused indole [alk](#page-0-0)aloid family. For the $[3 + 2]$ IMCC of chain substrate 1m, $Sc(OTf)$ ₃ failed, while $BF_3·Et_2O$ could realize the goal well. The scope of the carbocyclic ring size could be expanded to cyclononane. $[3+2]$ IMCC of substrates 1n and 1o gave cycloadducts 2n and 2o, respectively, in moderate yields. However, construction of a cyclodecane substructure had failed.

Figure 2. Lewis acid catalyzed $[3 + 2]$ IMCC of CoACP 1,1-diesters. Key: (a) Sc(OTf)₃ (0.4 equiv); (b) $BF_3·Et_2O$ (3.0 equiv) was used instead of Sc(OTf)₃, in DCM (2.5 mL), reflux. DCM = dichloromethane.

Removal of cobalt with "Bu₃SnH afforded the corresponding alkene products $(3-17)$ (Scheme 2).¹²

The successful construction of the medium-sized carbocyclic skeletons, as well as the bridged oxa-[n.2.1] skeletons, promises to provide an efficient and general strategy in application to synthesis of a larger family of biologically important natural products (Figure 1).

To explore the potential of this strategy further, several simple and target-orient[ed](#page-0-0) postfunctionalizations of 2a were carried out (Scheme 3). The cobalt-alkyne complex has rich chemistries, 10 including various cyclizations¹⁴ and cycloadditions¹⁵ for construct[io](#page-2-0)n of additional rings. Pauson−Khand [2 + 2 + [1\]](#page-3-0) $cycloaddition¹⁶$ of 2a with nor[bo](#page-3-0)rnene gave cycloadd[uct](#page-3-0) 18 $(18a/18b = 3:1)$. The introduction of a cyclopentane ring might find its appl[ica](#page-3-0)tion in construction of the 5−8 bicyclic core skeleton in natural products (Figure 1). A Diels−Alder [4 + 2] cycloaddition with 2,3-dimethyl-1,4-diene introduced another benzene ring and gave 6−8−6 tricycli[c](#page-0-0) product 19, which can be applied as a potential method for synthesis of kadsulignan family (Figure 1).^{5e} A furandione ring was also successfully introduced

Scheme 3. Several Convenient and Target-Oriented Postfunctionalizations of 2a

to afford 20, which supplied a potential application in construction of the bicyclic cyclooctane-furanone skeleton in natural products (Figure 1). Simmons−Smith cyclopropanation of 3 gave tricyclic product 21 with excellent stereoselectivity. The 3−8 bicyclic skeleton als[o](#page-0-0) exists in natural products (Figure 1).

In conclusion, we have reported the first intramolecular cycloaddition of CoACP. This constituted a general and effici[en](#page-0-0)t strategy for construction of medium-sized (cyclooctanoids and cyclononanoids) and related polycyclic skeletons. Features of this strategy include easy preparation of structurally diverse substrates, mild conditions, rich skeletal and substituent diversities, the inherent excellent stereochemistry of the bridged rings, and the rich and efficient postfunctionalizations (especially the introduction of additional rings). These are quite important in natural products synthesis and construction of structurally diverse libraries for lead discovery and chemical biology studies. Future investigations include the application of the strategy to total synthesis of natural products containing polycyclic skeletons and biological assay of the built compound library for lead discovery.

■ ASSOCIATED CONTENT

S Supporting Information

Experimental procedures and characterization data, including Xray crystal structures of products 2a, 3, and 18a (CIF). This material is available free of charge via the Internet at http://pubs. acs.org.

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

(1) (a) Blunt, J. W.; Copp, B. R.; Keyzers, R. A.; Munro, M. H. G.; Prinsep, M. R. Nat. Prod. Rep. 2014, 31, 160. and related series. (b) Fraga, B. M. Nat. Prod. Rep. 2013, 30, 1226. and related series. (c) Sun, H.; Li, S. Diterpenoid Chemistry; Chemical Industry Press: Beijing, 2011. (d) Hanson, J. R. Nat. Prod. Rep. 2009, 26, 1156 and related series. (e) Shi, Y. P. Mono and Sesquiterpenoid Chemistry; Chemical Industry Press: Beijing, 2008. (f) Wang, F. P. Alkaloid Chemistry; Chemical Industry Press: Beijing, 2008. (g) Breitmaier, E. Terpenes: Flavors, Fragrances, Pharmaca, Pheromones; Wiley-VCH: Weinheim, Germany, 2006. (h) Sell, C. S. A Fragrant Introduction to Terpenoid Chemistry; The Royal Society of Chemistry: Cambridge, U.K., 2003.

(2) (a) Galli, C.; Mandolini, L. Eur. J. Org. Chem. 2000, 3117. (b) Illuminati, G.; Mandolini, L. Acc. Chem. Res. 1981, 14, 95. (c) Winnik, M. A. Chem. Rev. 1981, 81, 491.

(3) For general reviews on construction of medium-sized carbocyclic skeletons: (a) Kotha, S.; Dipak, M. K. Tetrahedron 2012, 68, 397. (b) Inagaki, F.; Kitagaki, S.; Mukai, C. Synlett 2011, 594. (c) Majumdar, K. C.; Chattopadhyay, B. Curr. Org. Chem. 2009, 13, 731. (d) Nicolaou, K. C.; Bulger, P. G.; Sarlah, D. Angew. Chem., Int. Ed. 2005, 44, 4490. (e) Schrock, R. R.; Hoveyda, A. H. Angew. Chem., Int. Ed. 2003, 42, 4592. (f) Bear, B. R.; Sparks, S. M.; Shea, K. J. Angew. Chem., Int. Ed. 2001, 40, 820. (g) Maier, M. E. Angew. Chem., Int. Ed. 2000, 39, 2073. (h) Yet, L. Chem. Rev. 2000, 100, 2963. (i) Yet, L. Tetrahedron 1999, 55, 9349. (j) Molander, G. A. Acc. Chem. Res. 1998, 31, 603. (k) Booker-Milburn, K. I.; Sharpe, A. J. Chem. Soc., Perkin. Trans.1 1998, 983. (l) Harmata, M. Tetrahedron 1997, 53, 6235. (m) Ojima, I.; Tzamarioudaki, M.; Li, Z.; Donovan, R. J. Chem. Rev. 1996, 96, 635. (n) Molander, G. A.; Harris, C. R. Chem. Rev. 1996, 96, 307. (o) Griesbeck, A. G.; Henz, A.; Hirt, J. Synthesis 1996, 1261. (p) Tochtermann, W.; Kraft, P. Synlett 1996, 1029. (q) Roxburgh, C. J. Tetrahedron 1993, 49, 10749. (r) Zhao, R. B.; Wu, Y. L. Chin. J. Org. Chem. 1988, 8, 97.

(4) For general reviews on construction of medium-sized heterocyclic skeletons: (a) Sharma, A.; Appukkuttan, P.; Van der Eycken, E. Chem. Commun. 2012, 48, 1623. (b) Kleinke, A. S.; Webb, D.; Jamison, T. F. Tetrahedron 2012, 68, 6999. (c) Szostak, M.; Aube, J. Org. Biomol. Chem. 2011, 9, 27. (d) Chattopadhyay, S. K.; Karmakar, S.; Biswas, T.; Majumdar, K. C.; Rahaman, H.; Roy, B. Tetrahedron 2007, 63, 3919. (e) Shiina, I. Chem. Rev. 2007, 107, 239. (f) Clark, J. S. Chem. Commun. 2006, 3571. (g) Ghosh, S.; Sarkar, N. J. Chem. Sci. 2006, 118, 223. (h) Blankenstein, J.; Zhu, J. Eur. J. Org. Chem. 2005, 1949. (i) Nakamura, I.; Yamamoto, Y. Chem. Rev. 2004, 104, 2127. (j) McReynolds, M. D.; Dougherty, J. M.; Hanson, P. R. Chem. Rev. 2004, 104, 2239. (k) Dieters, A.; Martin, S. F. Chem. Rev. 2004, 104, 2199. (l) Schrock, R. R.; Hoveyda, A. H. Angew. Chem., Int. Ed. 2003, 42, 4592. (m) Nubbemeyer, U. Eur. J. Org. Chem. 2001, 1801. (n) Nubbemeyer, U. Top. Curr. Chem. 2001, 216, 125. (o) Kotha, S.; Sreenivasachary, N. Indian J. Chem. 2001, 40B, 763. (p) Furstner, A. Angew. Chem., Int. Ed. 2000, 39, 3012. (q) Maier, M. E. Angew. Chem., Int. Ed. 2000, 39, 2073. (r) Grubbs, R. H.; Chang, S. Tetrahedron 1998, 54, 4413. (s) Armstrong, S. K. J. Chem. Soc., Perkin Trans. 1 1998, 371. (t) Schuster, M.; Blechert, S. Angew. Chem., Int. Ed. 1997, 36, 2036. (u) Grubbs, R. H.; Miller, S. J.; Fu, G. C. Acc. Chem. Res. 1995, 28, 446. (v) Schmalz, H. G. Angew. Chem., Int. Ed. 1995, 34, 1833. (w) Rousseau, G. Tetrahedron 1995, 51, 2777. (x) Meng, Q.; Hesse, M. Top. Curr. Chem. 1992, 161, 107. (y) Evans, P. A.; Holmes, A. B. Tetrahedron 1991, 47, 9131.

(5) For reviews on construction of cyclooctanoid skeletons: (a) Jiao, L.; Yu, Z.-X. J. Org. Chem. 2013, 78, 6842. (b) Yu, Z.-X.; Wang, Y.; Wang, Y. *Chem.—Asian J.* **2010**, 5, 1072. (c) Saadi, J.; Brüdgam, I.; Reissig, H.-U. Beilstein J. Org. Chem. 2010, 6, 1229. (d) Michaut, A.; Rodriguez, J. Angew. Chem., Int. Ed. 2006, 45, 5740. (e) Chang, J.; Reiner, J.; Xie, J. Chem. Rev. 2005, 105, 4581. (f) Berndt, M.; Gross, S.; Holemann, A.; Reissig, H.-U. Synlett 2004, 422. (g) Murakami, M. Angew. Chem., Int. Ed. 2003, 42, 718. (h) Butkus, E. Synlett 2001, 1827. (i) Harmata, M. Acc. Chem. Res. 2001, 34, 595. (j) Li, Z.; Wang, F.; Chen, D. Chin. J. Org. Chem. 2000, 20, 282. (k) Mehta, G.; Singh, V. Chem. Rev. 1999, 99, 881. (l) Sieburth, S. M.; Cunard, N. T. Tetrahedron 1996, 52, 6251. (m) Petasis, N. A.; Patane, M. A. Tetrahedron 1992, 48, 5757. (n) Peters, J. A. Synthesis 1979, 321.

(6) For general reviews on medium-sized carbocyclic bridged skeletons: (a) Ruiz, M.; Lopez-Alvarado, P.; Giorgi, G.; Menendez, J. C. Chem. Soc. Rev. 2011, 40, 3445. (b) Zhao, W. Chem. Rev. 2010, 110, 1706. (c) Lopez, F.; Mascarenas, J. L. Chem.-Eur. J. 2007, 13, 2172. (d) Rigby, J. H. Tetrahedron 1999, 55, 4521. (e) Rigby, J. H. Acc. Chem. Res. 1993, 26, 579.

(7) Selected reviews on donor−acceptor cyclopropanes: (a) Cavitt, M. A.; Phun, L. H.; France, S. Chem. Soc. Rev. 2014, 43, 804. (b) Schneider, T. F.; Kaschel, J.; Werz, D. B. Angew. Chem., Int. Ed. 2014, 53, 5504. (c) Wang, Y.; Yin, J. J. Chem. Reagents 2013, 35, 318. (d) Wang, Z. Synlett 2012, 2311. (e) Tang, P.; Qin, Y. Synthesis 2012, 2969. (f) Mel'nikov, M. Y.; Budynina, E. M.; Ivanova, O. A.; Trushkov, I. V. Mendeleev Commun. 2011, 21, 293. (g) Campbell, M. J.; Johnson, J. S.; Parsons, A. T.; Pohlhaus, P. D.; Sanders, S. D. J. Org. Chem. 2010, 75, 6317. (h) Lebold, T. P.; Kerr, M. A. Pure Appl. Chem. 2010, 82, 1797. (i) Carson, C. A.; Kerr, M. A. Chem. Soc. Rev. 2009, 38, 3051. (j) De Simone, F.; Waser, J. Synthesis 2009, 3353. (k) Agrawal, D.; Yadav, V. K. Chem. Commun. 2008, 6471. (l) Yu, M.; Pagenkopf, B. L. Tetrahedron 2005, 61, 321. (m) Zimmer, R.; Ressig, H. U. Chem. Rev. 2003, 103, 1151. (n) Kulinkovich, O. G. Russ. Chem. Rev. 1993, 62, 839. (o) Wong, H. N. C.; Hon, M.-Y.; Tse, C.-W.; Yip, Y.; Tanko, J.; Hudlicky, T. Chem. Rev. 1989, 89, 165. (p) Wenkert, E. Acc. Chem. Res. 1980, 13, 27. (q) Danishefsky, S. Acc. Chem. Res. 1979, 21, 66.

(8) (a) Zhu, W.; Fang, J.; Liu, Y.; Ren, J.; Wang, Z. Angew. Chem., Int. Ed. 2013, 52, 2032. (b) Wang, Z.; Ren, J.; Wang, Z. Org. Lett. 2013, 15, 5682. (c) Bai, Y.; Tao, W.; Ren, J.; Wang, Z. Angew. Chem., Int. Ed. 2012, 51, 4112. (d) Xing, S.; Li, Y.; Li, Z.; Liu, C.; Ren, J.; Wang, Z. Angew. Chem., Int. Ed. 2011, 50, 12605. (e) Xing, S.; Pan, W.; Liu, C.; Ren, J.; Wang, Z. Angew. Chem., Int. Ed. 2010, 49, 3215. (f) Hu, B.; Xing, S.; Ren, J.; Wang, Z. Tetrahedron 2010, 66, 5671.

(9) (a) Christie, S. D. R.; Davoile, R. J.; Elsegood, M. R. J.; Fryatt, R.; Jones, R. C. F.; Pritchard, G. J. Chem. Commun. 2004, 2474. (b) Christie, S. D. R.; Davoile, R. J.; Jones, R. C. F. Org. Biomol. Chem. 2006, 4, 2683. (c) Lebold, T. P.; Carson, C. A.; Kerr, M. A. Synlett 2006, 364 A [3 + 2] cycloaddition of alkynylcyclopropane 1,1-diester with aldehyde or imine was reported. (d) Haubenreisser, S.; Hensenne, P.; Schroder, S.; Niggemann, M. Org. Lett. 2013, 15, 2262. (e) Miyake, Y.; Endo, S.; Moriyama, T.; Sakata, K.; Nishibayashi, Y. Angew. Chem., Int. Ed. 2013, 52, 1758.

(10) Reviews: (a) Green, J. R. Eur. J. Org. Chem. 2008, 6053. (b) Diaz, D. D.; Betancort, J. M.; Martin, V. S. Synlett 2007, 343. (c) Omae, I. Appl. Organomet. Chem. 2007, 21, 318. (d) Teobald, B. J. Tetrahedron 2002, 58, 4133. (e) Green, J. R. Curr. Org. Chem. 2001, 5, 809. (f) Welker, M. E. Curr. Org. Chem. 2001, 5, 785. (g) Fletcher, A. J.; Christie, S. D. R. J. Chem. Soc., Perkin Trans. 1 2000, 1657. (h) Nicholas, K. M. Acc. Chem. Res. 1987, 20, 207.

(11) CCDC 1008071 (2a), CCDC 1008070 (3), and CCDC 1008072 (18a) contain the supplementary crystallographic data for this letter. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/ cif.

(12) Because the cobalt−alkyne complex 2 could not give better NMR spectra, the cobalt moieties were removed to give the corresponding alkenes 3−17 and spectra of which were obtained (see Supporting Information).

(13) (a) Lian, J.-J.; Lin, C.-C.; Chang, H.-K.; Chen, P.-C.; Liu, R.-S. J. Am. Chem. Soc. 2006, 128, 9661. (b) Oh, C. H.; Kim, J. H.[;](#page-2-0) [Oh,](#page-2-0) [B.](#page-2-0) [K.;](#page-2-0) [Park,](#page-2-0) [J.](#page-2-0) [R.;](#page-2-0) [L](#page-2-0)ee, J. H. Chem.—Eur. J. 2013, 19, 2592. (c) Peng, Y.; Yu, M.; Zhang, L. Org. Lett. 2008, 10, 5187. (d) Lu, S.-C.; Wei, S.-C.; Wang, W.- X.; Zhang, W.; Tu, Z.-F. Eur. J. Org. Chem. 2011, 5905. (e) Biswas, S.; Batra, S. Adv. Synth. Catal. 2011, 353, 2861.

(14) (a) Djurdjevic, S.; Yang, F.; Green, J. R. J. Org. Chem. 2010, 75, 8241. (b) Closser, K. D. M.; Quintal, M.; Shea, K. M. J. Org. Chem. 2009, 74, 3680. (c) Onuki, K.; Asano, K.; Miyashita, M.; Nakamura, T.; Takahashi, Y.; Tanino, K.; Kuwajima, I. J. Am. Chem. Soc. 2003, 125, 1498. (d) Hosokawa, S.; Isobe, M. J. Org. Chem. 1999, 64, 37. (e) Jamison, T. F.; Shambayati, S.; Crowe, W. E.; Schreiber, S. L. J. Am. Chem. Soc. 1997, 119, 4353. (f) Nakamura, T.; Matsui, T.; Tanino, K.; Kuwajima, I. J. Org. Chem. 1997, 62, 3032. (g) Jamison, T. F.; Shambayati, S.; Crowe, W. E.; Schreiber, S. L. J. Am. Chem. Soc. 1994, 116, 5505. (h) Sammaikia, T.; Crowe, W. E.; Schreiber, S. L. J. Am. Chem. Soc. 1986, 108, 3128.

(15) (a) Dota, K.; Shimizu, T.; Hasegawa, S.; Miyashita, M.; Tanino, K. Tetrahedron Lett. 2011, 52, 910. (b) Mitachi, K.; Shimizu, T.; Miyashita, M.; Tanino, K. Tetrahedron Lett. 2010, 51, 3983. (c) Iwasawa, N.; Inaba, K.; Nakayama, S.; Aoki, M. Angew. Chem., Int. Ed. 2005, 44, 7447. (d) Kondo, F.; Shimizu, T.; Tanino, K.; Miyashita, M. Org. Lett. 2002, 4, 2217. (e) Sakurada, F.; Iwamoto, M.; Iwasawa, N.Org. Lett. 2000, 2, 871. (f) Patel, M. M.; Green, J. R. Chem. Commun. 1999, 509.

(16) Reviews: (a) Lee, H.-W.; Kwong, F.-Y. Eur. J. Org. Chem. 2010, 789. (b) Shibata, T. Adv. Synth. Catal. 2006, 348, 2328. (c) Mainolfi, N.; Gibson, S. E. Angew. Chem., Int. Ed. 2005, 44, 3022. (d) Urgoiti, J. B.; Anorbe, L.; Serrano, L. P.; Dominguez, G.; Castells, J. P. Chem. Soc. Rev. 2004, 33, 32. (e) Bonaga, L. V. R.; Krafft, M. E. Tetrahedron 2004, 60, 9795. (f) Gibson, S. E.; Stevenazzi, A. Angew. Chem., Int. Ed. 2003, 42, 1800. (g) Xie, X.; Yang, G.; Zhao, G. Chin. J. Org. Chem. 2002, 22, 610. (h) Brummond, K. M.; Kent, J. L. Tetrahedron 2000, 56, 3263. (i) Geis, O.; Schmalz, H. Angew. Chem., Int. Ed. 1998, 37, 911. (j) Schore, N. E. Org. React. 1991, 40, 1. (k) Schore, N. E. Chem. Rev. 1988, 88, 1081.

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Scheme 2 and Figure 2 were corrected on January 16, 2015.